

Reg. No.

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B.E / B.TECH. DEGREE EXAMINATIONS, MAY 2024

Second Semester

CY18251 – ORGANIC CHEMISTRY*(Chemical Engineering)***Regulation 2018/2018A****TIME: 3 HOURS****MAX. MARKS: 100**

COURSE OUTCOMES	STATEMENT	RBT LEVEL
CO1	Enable the students to learn the various reaction mechanisms.	2
CO2	Familiarize the industrially important catalysts.	2
CO3	The synthesise of different types of dyes.	3
CO4	Impart knowledge on the synthetic utility of organic reagents.	3
CO5	Understanding of thermal method of analysis and chromatographic techniques.	3

PART- A (10 x 2 = 20 Marks)

(Answer all Questions)

	CO	RBT LEVEL
1. Mention any two examples for free radicals.	1	1
2. Highlight the role of the reagent N-Bromosuccinimide.	1	2
3. Distinguish the homogeneous and heterogeneous catalysis.	2	2
4. Draw the structure of Wilkinson Catalyst.	2	2
5. List any two requisites of a dye.	3	1
6. What is a mordant dye?	3	1
7. Write the importance of a synthon.	4	2
8. What is a multistep synthesis?	4	2
9. List any two eluents used in column chromatography.	5	1
10. Mention any two applications of paper chromatography.	5	1

PART- B (5 x 14 = 70 Marks)

	Marks	CO	RBT LEVEL
11.(a) (i) Propose a probable mechanism for Friedel Craft alkylation.	(7)	1	2
(ii) Provide a feasible mechanism for Benzoin condensation.	(7)	1	2

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(OR)

- (b) (i) Write a note on Aldol condensation. (7) 1 2
 (ii) Propose the mechanism for Pinacol rearrangement. (7) 1 2

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- 12.(a) (i) Explain the epoxidation reaction with an example. (7) 2 2
 (ii) Summarize the role of sonocatalysts in hydrogenation reactions. (7) 2 2

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(OR)

- (b) Describe the Shell process and list any two industrial applications. (14) 2 2

- 13.(a) (i) Outline the Witt's theory of colour and constitution. (7) 3 2
 (ii) Interpret the synthesis and uses of the Malachite green dye. (7) 3 2

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(OR)

- (b) (i) Discuss the synthesis and uses of Methyl red. (7) 3 2
 (ii) Explain the synthesis and uses of the Fluorescein dye. (7) 3 2

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- 14.(a) Explain any four synthetic utility of Grignard reagent. (14) 4 3

(OR)

- (b) Make use of diethyl malonate in any four synthetic utility. (14) 4 3

- 15.(a) (i) Apply the thermo gravimetric principle in the decomposition of copper sulphate pentahydrate. (7) 5 3

- (ii) Construct and explain the analytical method Electrophoresis. (7) 5 3

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(OR)

- (b) Utilize the principle and instrumentation of High Pressure Liquid Chromatography (HPLC) in separation of analytes. (14) 5 3

PART- C (1 x 10 = 10 Marks)
(Q.No.16 is compulsory)

	Marks	CO	RBT LEVEL
16. Apply the principle and instrumentation of gas chromatography in separation of analytes.	(10)	5	3