

Reg. No.

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**B.E / B.TECH. DEGREE EXAMINATIONS, MAY 2024**

Third Semester

**BT18301 – BIOORGANIC CHEMISTRY***(Biotechnology)***(Regulation 2018/2018A)****TIME: 3 HOURS****MAX. MARKS: 100**

<b>CO 1</b>	To know in detail about the elements of atom, charges and their bonding rule	<b>2</b>
<b>CO 2</b>	To understand the various kinetic properties and types of reaction mechanisms	<b>2</b>
<b>CO 3</b>	To understand the possible bio-organic reactions involved in biosynthesis	<b>2</b>
<b>CO 4</b>	To analyze various bioorganic based productions	<b>4</b>
<b>CO 5</b>	To apply the concepts of Bioorganic reactions	<b>3</b>

**PART- A (10 x 2 = 20 Marks)**

(Answer all Questions)

	CO	RBT LEVEL
1. Outline on how Electronegativity influence the course of reaction.	1	2
2. Explain acid base equilibria in resonance reactions.	1	1
3. Illustrate steric effects with example.	2	2
4. Summarize on nucleophiles in substitution reactions.	2	2
5. Explain microscopic reversibility in one step and two step reaction.	3	2
6. Give the role Transition state theory with respect to life of intermediate.	3	1
7. What is British anti Lewis?	4	4
8. Why coenzymes are required for catalytic activity?	4	4
9. Show alkyl group transfer with an example.	5	3
10. How C-C bond formation and fission occurs in reactions of bioorganic chemistry?	5	3

**PART- B (5 x 14 = 70 Marks)**

	Marks	CO	RBT LEVEL
11. (a) Experiment with the principle of staggered and eclipse for conformations analysis of butane and cycloalkane.	14	1	2
<b>(OR)</b>			
(b) Construct how energies undergo sort of intermixing in SP <sup>3</sup> hybridization to form hybrid orbitals.	14	1	2

12. (a) Explain how Nucleophilic addition on aldehyde and ketone results in acetals and ketals formation. 14 2 2
- (OR)
- (b) Describe the possibilities of inversion and retention of configuration in  $S_N1$  and  $S_N2$  reactions. 14 2 2
13. How  $\Delta G$ ,  $\Delta S$ ,  $\Delta H$  helps to know the Kinetic, thermodynamic reversibility, and thermodynamics of coupled reactions. 14 3 2
- (OR)
- (b) Make use of rate law and rate constant for sequential reaction to trap the intermediate. 14 3 2
14. (a) Simplify on catalysis by organized aggregates and phases. 14 4 4
- (OR)
- (b) Infer on mechanistic aspects of Intra molecular reactions. 14 4 4
15. (a) How the template DNA sequence can be known with chain termination method of DNA sequencing and report the sequence. 14 5 3
- (OR)
- (b) Explain the synthesis of synthetic peptide by solid phase peptide synthesis. 14 5 3

**PART- C (1 x 10 = 10 Marks)**

(Q.No.16 is compulsory)

- |  | Marks | CO | RBT<br>LEVEL |
|--|-------|----|--------------|
| 16. Assess the stereochemical activity around the tetrahedral carbon in various bioorganic compounds or elements | 10    | 1  | 4            |

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